Matter.notebook October 15, 2013

## What is matter?

**Matter** is anything that has mass and takes up space (volume).

Mass--the amount of matter

- --measured using a scale
- --reported in grams
- --always constant despite location

**Weight**--a measure of the gravitational force exerted on an object

- --measured using a spring scale
- --reported in Newtons or pounds
- --not constant

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Physical Change -- a change that

does not change the identity of a substance. Can be undone.

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What are the signs of a chemical change?

Color change

Gas formation (bubbles/fizzing)

Light Production

Temperature Change

Formation of a new solid

What properties define matter?

**Physical**—can be observed or measured without changing the identity of the substance.

Color

Odor

State/Phase

Density

Malleability

Ductility

Solubility

Conductivity (Thermal vs Electrical)

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Chemical Properties—properties that define matter's ability to change into matter with different properties.

Flammability

Reactivity (chemical reactions)

Chemical Change—the atoms of a substance are rearranged into substances that have new and different properties. Typically not reversible.

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## Law of Conservation of Mass

The mass that you start with in a chemical or physical change is the mass that you will end with. Matter.notebook October 15, 2013

EQ. How can you tell if a physical or a chemical change is occurring?

Hypothesis: If a chemical change is taking place, then...

Set up data table:

Name of	Detailed	Physical or
Change	Observations	Chemical

Oct 14-7:59 AM